

AUDITORIUM



Potential consequences of chemicals not being properly managed:













Domestic and foreign university laboratory accidents



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Strategy:
Design and layout

Management: Chemicals management



Technology: Violation of operations

Case 1: 2018-12-26 Beijing Jiaotong University waste explosion Three graduate students died

Case 2: A female researcher dies at the University of California laboratory explosion in 2015

Case 3: 2015-12-18 Tsinghua University explosion caused a postdoctoral death and burned three laboratories

Case 4: 2016-3-16 The University of Hawaii laboratory explosion caused a loss of one arm and the school was fined a million fines.

Domestic and foreign university laboratory accidents



At 9:34 am on December 26, 2018, an explosion occurred in a laboratory at Building 2 of the East Campus of Beijing Jiaotong University. The accident caused a total of three students who participated in the experiment to die, and the site was over 60

square meters.



Background: Lab Incidents in Universities of China

| No. | Incident time | Incident brief introduction | Result |
|-----|------------------------|--|--|
| 1. | 2019-03-27 19:00 | A lab explosion happened in Fudan University of Shanghai | 1 student got serious injury and lab got burned |
| 2. | 2018-12-26 09:30 | An explosion occurred in the environmental engineering lab of Beijing Jiaotong University during handling waste | This case caused 3 graduate students' fatality |
| 3. | 2018-11-11 10:00 | An explosion occurred in a lab of Hanlin College of Nanjing University of Traditional Chinese Medicine | Several students got burned |
| 4. | 2016-09-21 | An explosion happened in a chemical and biological lab at the Songjiang campus of Donghua University | This case caused 3 graduate students injury |
| 5. | 2016-1-10 12:00 | A fire broke out in a lab in the science and technology building of Beijing University of Chemical Technology | The lab got burned |
| 6. | 2015-12-18 10:00 | An explosion occurred in the chemistry lab of Tsinghua University | One doctoral candidate died in this case |
| 7. | 2015-06-17 16:30 | Lithium exploded in a lab on the 2nd floor of the physics building of Suzhou University | The lab got burned |
| 8. | 2015-04-05 noontime | An explosion hit a lab at the College of Chemical Engineering, China University of Mining and Technology in Xuzhou of Jiangsu province | This case caused 1 dead and 5 seriously injured |
| 9. | 2014-12-04 11:00 | A lab explosion at Changzhou Institute of Engineering in Jiangsu province | The lab got burned |
| 10. | 2013-04-30 09:00 | An explosion occurred in an abandoned laboratory at Nanjing University of Science and Technology during demolition work | This case caused 1 dead and 3 seriously injured |
| 11. | 2012-6-3 07:30 | A fire broke out in a lab at Peking University | Two research labs got burned |
| 12. | 2012-3-6 18:00 | A fire broke out in a lab building of Peking University medical center | One lab got burned |
| 13. | 2011-12-07 11:00 | Tianjin Nankai University chemical lab accident | One female student suffered from serious hand injury |
| 14. | 2011-04-14 15:45 | An explosion occurred in a lab of the Chemical Engineering Institute of Sichuan University at Jiangan campus | 3 students got injured |
| 15. | 2009-10-23 13:00 | An explosion happened in a lab of new building 5 at Beijing Institute of Technology) | The lab got burned, 5 students got injured |
| 16. | P22008-07-11 10:00 | An explosion happened at lab 510, Yunnan Institute of Microbiology, Beiyuan, Yunnan University | One doctoral student got seriously injured |

Current status of chemicals management in colleges and universities



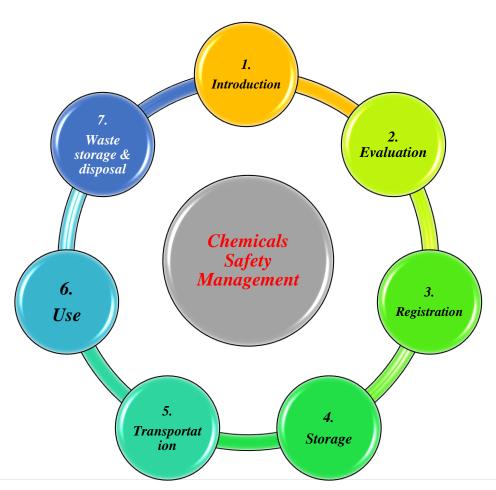
- 1. The management rules and regulations are not perfect;
- 2. The control of the procurement process is not strict;
- 3. The storage management is relatively random and lacks rationality;
- 4. The warehouse does not meet the safety design standards;
- 5. Violation of lab safety guidelines.
- 6. Insufficient supply of emergency equipment;
- 7. Safety inspection and management are not in place;
- 8. Experimental staff's safety awareness is weak.

Countermeasures for Chemicals Management in Colleges and Universities



- 1. Optimize the management mechanism and improve the rules and regulations and operating procedures for chemicals management;
- 2. Establish a group for chemical safety management to implement safety supervision responsibilities;
- 3. Strengthen safety education and training;
- 4. Establish a safety management system for the entire life cycle of chemicals using information technology.

Chemicals Safety Management——Full life cycle management:





Introduction, evaluation and registration of chemicals:

- Need to determine the applicant, use and place of use of the chemical;
- Relevant factors such as the type of chemical, storage conditions, operational and environmental requirements, PPE, emergency treatment, etc.
- Registered in the unit management system for the above information .





Storage and internal transportation:

- Special requirements for the storage and transport of regulatory chemicals
- General chemicals' storage and transportation requirements

Storage area:

- Central warehouse (maximum storage quantity)
- Laboratory (total 100L or 100Kg, individually packaged 20L/25Kg)







Use:

- Laboratory engineering control
- Lab safety guidelines
- PPE

Emergency treatment:

- Equipment preparation
- Response training







Waste disposal:

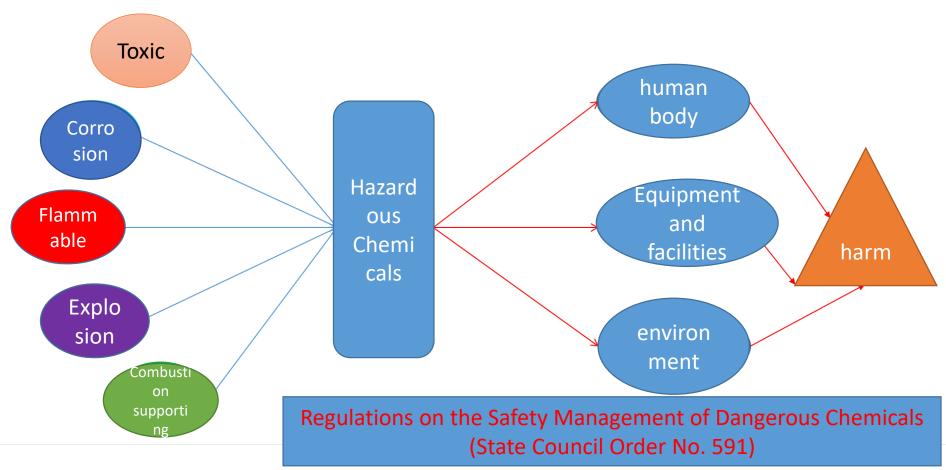
- The classification of experimental waste should be based on relevant national waste classification regulations.
- 2. Local waste disposal requirements
- 3. The extent to which the school can pretreat



Key to chemicals safety management:

- 1. Hazardous chemicals (including highly toxic chemicals)
- 2. Precursor chemicals
- 3. Explosive chemicals
- 4. Regulatory chemicals

1. Hazardous chemicals (definition)



1.Hazardous chemicals (including highly toxic chemicals)

There are 2,828 kinds of dangerous chemicals listed in the *Catalogue of Hazardous Chemicals* (2015 Edition), including 148 highly toxic chemicals.

Highly toxic chemicals - generally chemicals with severe toxic hazards, including synthetic chemicals and their mixtures and natural toxins, as well as chemicals that are acutely toxic and pose a public safety hazard.

The highly toxic chemicals should be stored separately in a dedicated warehouse, and a two-person transfer and double-custodian system should be implemented. The storage unit shall report the storage quantity, storage location and the status of the management personnel to the local safety supervision department and the public safety organ for the record.

2.Precursor chemicals - refer to substances regulated by the state which can be used to manufacture drugs, including precursor chemicals, raw materials and chemical additives. Identification basis: Catalogue of Classification and Varieties of Precursor Chemicals and Supplementary Catalogue Announced by the State Council. Regulatory requirements: Regulations on the Administration of Precursor Chemicals (Order No. 445 of the State Council, revised in 2018). **Precursor chemicals among the most common hazardous chemicals:** toluene, acetone, methyl ethyl ketone, potassium permanganate, sulfuric acid, hydrochloric acid, and bromine.

There are 32 substances in 3 categories of precursor chemicals

Category I

- 1.1-phenyl-2-propanone
- 2.3,4-methylenedioxyphenyl-2-propanone
- 3. Piperonal
- 4. Jaundice
- 5. Astragalus oil
- 6. Isoflavone
- 7. N-acetylanthranilic acid
- 8. Anthranilic acid
- 9. Lycopene
- 10. Ergotamine
- 11. Ergometrine
- 12. Ephedrine, pseudoephedrine, racemic ephedrine and other ephedrine substances
- 13. Hydroxyimine
- 14.1-phenyl-2-bromo-1-propanone
- 15.3-oxo-2-phenylbutyronitrile
- 16. O-chlorophenylcyclopentanone
- 17. N-phenethyl-4-piperidone
- 18. 4-anilino-N-phenethylpiperidine
- 19. N-methyl-1-phenyl-1-chloro-2-propylamine

Category II

- 1. Phenylacetic acid
- 2. Acetic anhydride
- 3. Chloromethane
- 4. Ether
- 5. Piperidine
- 6. Bromine[3]
- 7.1-phenyl-1-acetone

Category III

- 1. Toluene
- 2. acetone
- 3. Methyl ethyl ketone
- 4. potassium permanganate
- 5. sulfuric acid
- 6. hydrochloric acid

3.Chemicals can be used to produce explosives - can be used as raw materials or accessories to make explosive properties.

Identification basis: List of Explosive Dangerous Chemicals (2017 Edition)

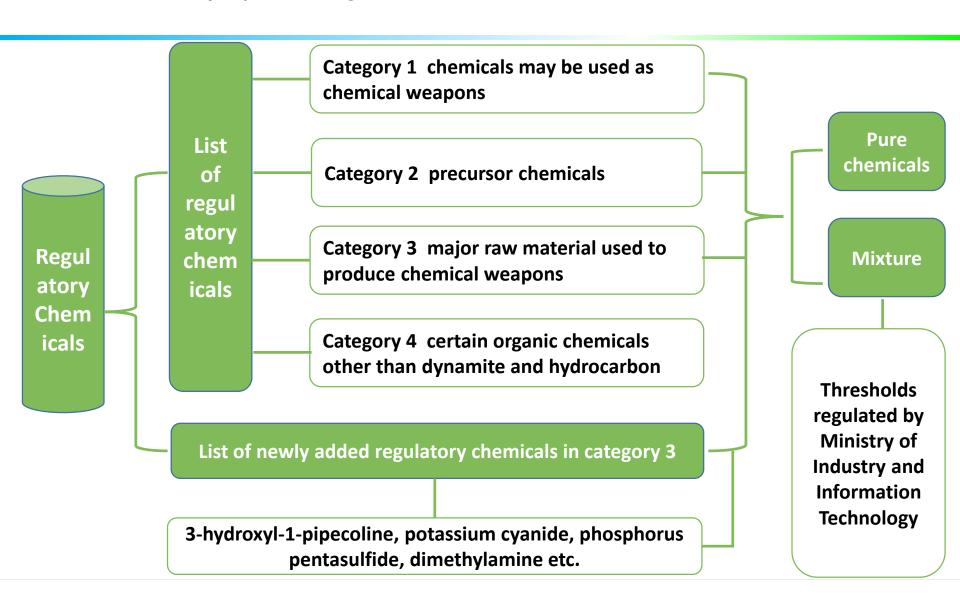
Regulatory requirements: Administrative Measures for the Safety of Explosive

Hazardous Chemicals (Order No. 154 of the Ministry of Public safety)

The units that are vulnerable to explosives and dangerous chemicals shall establish an information system for explosive chemicals that is easy to blast, and realize the interconnection and intercommunication with the information system of the public safety organs. Units that are vulnerable to blast dangerous chemicals shall implement electronic tracking and labeling management for the explosive chemicals that are easy to blast, and monitor and record the flow of dangerous chemicals that are easy to blast.

4. Regulatory chemicals - those items that are strictly controlled by the state and local governments and can be used to produce chemical weapons.

Identification basis and regulatory requirements: *Regulations of the People's Republic of China on the Control of Chemicals* (Decree No.190 of the State Council of the People's Republic of China)



SDS vs MSDS

SDS is the acronym for Safety Data Sheet. The REACH Regulation (Order No. EC 1907/2006) specifies the content of the SDS in Appendix II, which was revised in May 2010 by Directive 453/2010.

MSDS: MSDS is the acronym for Material Safety Data Sheet. There are many standards for the development of MSDS, mainly standards set by GHS, ANSI, ISO, OSHA, and WHMIS.

The European Union and the International Organization for Standardization (ISO) use the SDS terminology. However, in the United States, Canada, Australia and many countries in Asia, SDS (Safety Data Sheet) can also be used as MSDS (Material Safety Data Sheet), the role of two technical documents. Basically consistent.

In 2008, China revised the standard GBT_16483-2008 Safety Data Sheet for Chemical Products Contents and Order of Sections and unified it with the International Organization for Standardization, abbreviated as SDS.

The two acronyms SDS and MSDS play a completely identical role in the supply chain, with only minor differences in content.



SDS (Safety Data Sheet) contains 16 sections: 1. Chemical product and company information 9. Physical and chemical properties 2. Hazards identification 10. Stability and reactivity 3. Composition/information on ingredients 11. Toxicological information 12. Ecological information 4. First-aid measures 5. Fire-fighting measures 13. Disposal considerations 14. Transport information 6. Accidental release measures 7. Handling and storage 15. Regulatory information 8. Exposure controls/personal protection 16. Other information

- SDS content
- SECTION 1 Chemical Product & Company Information
- include:
 - Chemical name
 - Information such as the manufacturer's name, address, zip code, telephone number, emergency number, fax number, and email address.

SECTION 1: Identification of the substance/mixture and of the company/undertaking

1.1 Product identifiers

Product name : Acetone

Product Number : 650501

Brand : Sigma-Aldrich Index-No. : 606-001-00-8

REACH No. : 01-2119471330-49-XXXX

CAS-No. : 67-64-1

1.2 Relevant identified uses of the substance or mixture and uses advised against

Identified uses : Laboratory chemicals, Manufacture of substances

1.3 Details of the supplier of the safety data sheet

Company : Sigma-Aldrich Company Ltd.

The Old Brickyard

NEW ROAD, GILLINGHAM

Dorset SP8 4XT

UNITED KINGDOM

Telephone : +44 (0)1747 833000 Fax : +44 (0)1747 833313

E-mail address : eurtechserv@sial.com

1.4 Emergency telephone number

Emergency Phone # +44 (0)870 8200418 (CHEMTREC)

Call for help in the event of a chemical accident

- SECTION 2 Hazards Identification A brief overview of the most important hazards and effects of this chemical.
- Mainly include:
 - Hazard category,
 - Route of invasion,
 - Health hazard,
 - Environmental hazards,
 - Information on the danger of explosion.



SECTION 2: Hazards identification

2.1 Classification of the substance or mixture

Classification according to Regulation (EC) No 1272/2008

Flammable liquids (Category 2), H225

Eye irritation (Category 2), H319

Specific target organ toxicity - single exposure (Category 3), Central nervous system, H336

For the full text of the H-Statements mentioned in this Section, see Section 16.

Danger

2.2 Label elements

Labelling according Regulation (EC) No 1272/2008

Pictogram

Signal word

Hazard statement(s)

H225

H319

H336

Highly flammable liquid and vapour. Causes serious eye irritation.

May cause drowsiness or dizziness.

The hazard of chemicals can be preliminarily judged by pictograms, which are generally found on SDS, safety labels and chemical packaging.

| Precautionary statement(s) | |
|----------------------------|--|
| P210 | Keep away from heat, hot surfaces, sparks, open flames and other ignition sources. No smoking. |
| P233 | Keep container tightly closed. |
| P261 | Avoid breathing dust/ fume/ gas/ mist/ vapours/ spray. |
| P280 | Wear protective gloves/ protective clothing/ eye protection/ face protection. |
| P303 + P361 + P353 | IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water. |
| P370 + P378 | In case of fire: Use dry sand, dry chemical or alcohol-resistant foam to extinguish. |

2.3 Other hazards

EUH066

Supplemental Hazard information (EU)

This substance/mixture contains no components considered to be either persistent, bioaccumulative and toxic (PBT), or very persistent and very bioaccumulative (vPvB) at levels of 0.1% or higher.

Repeated exposure may cause skin dryness or cracking.

- SECTION 3 Composition/Information on Ingredients Indicate whether the chemical is a pure chemical or a mixture.
 - For pure chemicals, the chemical name or trade name and common name should be given.
 - For mixture, the concentration or concentration range of the hazardous component should be given .
 - No matter it is a pure chemical or a mixture, if it contains harmful components, a Chemical Abstracts Index registration number (CAS

number) should be given.

SECTION 3: Composition/information on ingredients

3.1 Substances

Formula : C₃H₆O

Molecular weight : 58.08 g/mol

CAS-No. : 67-64-1

EC-No. : 200-662-2

Index-No. : 606-001-00-8

Registration number : 01-2119471330-49-XXXX

Hazardous ingredients according to Regulation (EC) No 1272/2008

| Component | | Classification | Concentration | | | |
|---------------------|-----------------------|----------------------------------|---------------|--|--|--|
| Acetone | | | | | | |
| CAS-No. | 67-64-1 | Flam. Liq. 2; Eye Irrit. 2; STOT | <= 100 % | | | |
| EC-No. | 200-662-2 | SE 3; H225, H319, H336 | | | | |
| Index-No. | 606-001-00-8 | Concentration limits: | | | | |
| Registration number | 01-2119471330-49-XXXX | >= 20 %: STOT SE 3, H336; | | | | |

For the full text of the H-Statements mentioned in this Section, see Section 16.

- SECTION 4 First Aid Measures Refers to the treatment of on-site self-rescue or mutual rescue when the operator is accidentally injured, include:
 - eye contact,
 - skin contact.
 - Inhalation,
 - First aid measures for ingestion.
 - Three ways of contacting people exposed to such chemicals, by controlling contact to avoid injury.



SECTION 4: First aid measures

4.1 Description of first aid measures

General advice

Consult a physician. Show this safety data sheet to the doctor in attendance.

If inhaled

If breathed in, move person into fresh air. If not breathing, give artificial respiration. Consult a physician.

In case of skin contact

Wash off with soap and plenty of water. Consult a physician.

In case of eye contact

Rinse thoroughly with plenty of water for at least 15 minutes and consult a physician.

If swallowed

Do NOT induce vomiting. Never give anything by mouth to an unconscious person. Rinse mouth with water. Consult a physician.

4.2 Most important symptoms and effects, both acute and delayed

The most important known symptoms and effects are described in the labelling (see section 2.2) and/or in section 11

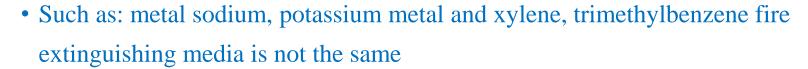
4.3 Indication of any immediate medical attention and special treatment needed

No data available

• SECTION 5 Fire Fighting Measures It mainly indicates the physical and chemical specific hazards of chemicals, and is suitable for fire extinguishing media, unsuitable fire extinguishing media, and personal protection of firefighters.

• include:

- Hazardous characteristics,
- Extinguishing media and methods,
- Fire fighting precautions, etc.





SECTION 5: Firefighting measures

5.1 Extinguishing media

Suitable extinguishing media

Dry powder Dry sand

Unsuitable extinguishing media

Do NOT use water jet.

5.2 Special hazards arising from the substance or mixture
No data available

5.3 Advice for firefighters

Wear self-contained breathing apparatus for firefighting if necessary.

5.4 Further information

Use water spray to cool unopened containers.

Respiratory equipment required to enter the fire.

Fire-fighting materials that

are required for fire, do not use spray water. Mist water

can be used to cool down.

• SECTION 6 Accidental Release Measures Refers to simple and effective emergency measures, precautions and elimination methods that can be used on site after chemical leakage.

• include:

- Emergency action,
- Emergency personnel protection,
- Environmental protection measures,
- Eliminate methods and other contents.





Emergency measures can be used as the content of the on-site disposal plan

SECTION 6: Accidental release measures

6.1 Personal precautions, protective equipment and emergency procedures

Use personal protective equipment. Avoid breathing vapours, mist or gas. Ensure adequate ventilation. Remove all sources of ignition. Evacuate personnel to safe areas. Beware of vapours accumulating to form explosive concentrations. Vapours can accumulate in low areas. For personal protection see section 8.

6.2 Environmental precautions

Prevent further leakage or spillage if safe to do so. Do not let product enter drains.

6.3 Methods and materials for containment and cleaning up

Contain spillage, and then collect with non-combustible <u>absorbent material</u>, (e.g. sand, earth, <u>diatomaceous earth</u>, <u>vermiculite</u>) and <u>place in container for disposal according to local</u> / national regulations (see section 13).

6.4 Reference to other sections

For disposal see section 13.

Adsorbent materials can be included in the emergency supplies list

• SECTION 7 Handling and Storage Mainly refers to the information on the handling and safe storage of chemicals.

- Safety precautions in handling;
- Safe storage conditions.



Operational requirements

SECTION 7: Handling and storage

7.1 Precautions for safe handling

Avoid contact with skin and eyes. Avoid inhalation of vapour or mist.

Keep away from sources of ignition - No smoking. Take measures to prevent the build up of electrostatic charge.

For precautions see section 2.2.

7.2 Conditions for safe storage, including any incompatibilities

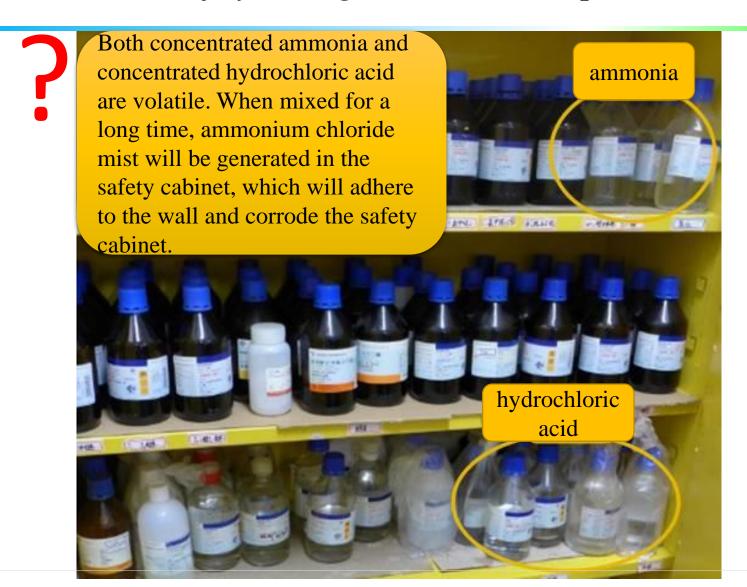
Keep container tightly closed in a dry and well-ventilated place. Containers which are opened must be carefully resealed and kept upright to prevent leakage. Store in cool place.

7.3 Specific end use(s)

Apart from the uses mentioned in section 1.2 no other specific uses are stipulated

Storage conditions

Chemicals Safety Management - common problem



Chemicals Safety Management - common problem



Problem: Volatile substances in the operating environment may be flammable gases, vapors or toxic and hazardous volatiles.

Recommendation: Volatile reagents should be placed in a cabinet with good ventilation and grounding. The seal of the reagent bottle should be checked after the experiment. Respiratory protective equipment should be used if necessary during the experiment.

Chemicals Safety Management - common problem





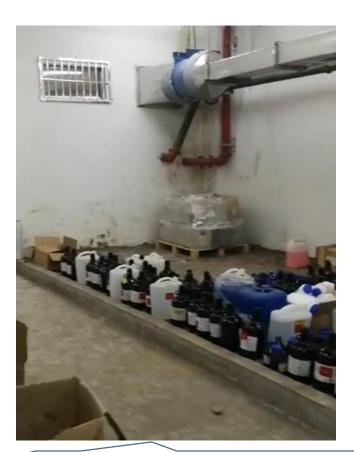
- Problem: no appropriate cabinets are used to store chemicals;
- Shelf deformation
- Chemicals are mixed.
- Recommendation: Use chemical-specific cabinets for storage;
- Reasonable storage according to the physical and chemical properties of the chemicals.

| 8.2 | Storage of Chemical Reagent in Laboratory | |
|-------|---|---|
| 8.2.1 | | Establish directory, MSDS or safety card for hazardous chemicals of the laboratories for easy reference; clean the overdue drugs regularly to avoid accumulation |
| 8.2.2 | | The storeroom, storage area and storage cabinet should be ventilated, thermal-insulated, sun-proof and safe; the storage area of organic solvents should stay away from the source of heat and ignition; the reagents subject to easy leakage and volatilization should be fully ventilated; no outlet or wiring board should exist in the reagent cabinet; the chemicals should be orderly placed by category; protection functions against secondary leakage, absorption or overflow should be provided; do not stack the reagents, have mixed storage of contraindicated chemicals, place solids and liquids in disorder or keep reagent bottle opened; do not store chemical reagents in experiment bench unless it has a baffle. |
| 8.2.3 | laboratories conform to the requirements of specification | In principle, the total quantity of hazardous chemicals should not exceed 100L or 100kg, among which, the total quantity of inflammable & explosive chemicals should not exceed 50L or 50kg, and capacity of single packaged container should not exceed 20L or 20kg; leakage alarm and ventilation linkage device must be installed if single experiment device has Class A material tanks over 10L, or Class B material tanks over 20L or Class C material tanks over 50L. The storage quantity of hazardous chemicals can be checked by laboratory area on basis of the standard laboratory area of 50 m ² . |
| 8.2.4 | | Up-to-standard labels should be pasted on packages of chemicals and be repasted if such chemicals are transferred or divided into other packages. The chemical labels should be timely supplemented if falling or becoming fuzzy and corroded; otherwise, it should be handled as wasted chemicals. |

| 8.4 | Management of Controlled Chemicals |
|-------|---|
| 8.4.1 | Check if precursor chemicals and potential explosives are stored by category, kept by a specially-assigned person and provided with receiving, use and handling records Regulatory chemicals should be managed based on "five doubles" systems (double-person acceptance, double-person keeping, double-person receiving, double locks and double accounts); specific safe box should be used for storing the highly toxic products, firmly fixed and managed based on double persons and double locks; the highly toxic products with high volatility and low flash point should be stored in explosion-proof refrigerator and provided with two locks; monitoring and alarm device should be used; the highly toxic products should be used by at least two operators, and be handled in strict accordance with the SOP. |
| 8.4.2 | Check if explosives are isolated, stored at limited amount, used and destroyed as specified Check the site and record book; annual list should be provided by functional departments by public safety department |
| 8.4.3 | Check if narcotic drugs and psychotropic drugs are stored in specific safe box and provided with standard accounts for receiving, use and handling Check the site and record book; annual list should be provided by functional departments |



Normal fridge for high volatility and low flash point chemicals



Five double:

- 1. double-person acceptance
- 2. double-person keeping
- 3. double-person receiving
- 4. double locks
- 5. double accounts

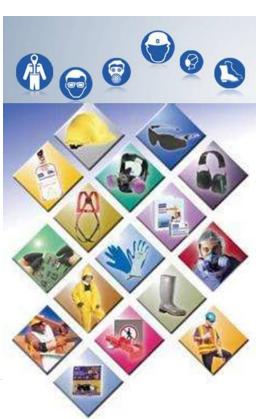




We already have the good condition, but how to manage

• SECTION 8 Exposure Controls/Personal Protection Protective methods and means to protect workers from chemical hazards during production, handling and use of chemicals.

- Maximum allowable concentration STEL, TWA
- Engineering control fume hood,
- Respiratory protection, eye protection, body protection, hand protection;
- Other protection requirements.



SECTION 8: Exposure controls/personal protection

8.1 Control parameters

Components with workplace control parameters

| Component | CAS-No. | ValueForm | Control | Basis |
|-----------|---------|-------------|-------------|---|
| | | of exposure | parameters | |
| Acetone | 67-64-1 | TWA | 500 ppm | Europe. Commission Directive |
| | | | 1,210 mg/m3 | 2000/39/EC establishing a first list of |
| | | | | indicative occupational exposure |
| | | | | limit values |
| | Remarks | Indicative | | |
| | / | TWA | 500 ppm | UK. EH40 WEL - Workplace |
| | | | 1,210 mg/m3 | Exposure Limits |
| | | STEL | 1,500 ppm | UK. EH40 WEL - Workplace |
| | | | 3,620 mg/m3 | Exposure Limits |

Predicted No Effect Concentration (PNEC)

| Compartment | Value |
|-------------------------------|------------|
| Soil | 33.3 mg/kg |
| Marine water | 1.06 mg/l |
| Fresh water | 10.6 mg/l |
| Marine sediment | 3.04 mg/kg |
| Fresh water sediment | 30.4 mg/kg |
| Onsite sewage treatment plant | 100 mg/l |

8.2 Exposure controls

Appropriate engineering controls

Handle in accordance with good industrial hygiene and safety practice. Wash hands before breaks and at the end of workday.

Personal protective equipment

Eye/face protection

Face shield and safety glasses Use equipment for eye protection tested and approved under appropriate government standards such as NIOSH (US) or EN 166(EU).

Skin protection

Handle with gloves. Gloves must be inspected prior to use. Use proper glove removal technique (without touching glove's outer surface) to avoid skin contact with this product. Dispose of contaminated gloves after use in accordance with applicable laws and good laboratory practices. Wash and dry hands.

The selected protective gloves have to satisfy the specifications of EU Directive 89/686/EEC and the standard EN 374 derived from it.

Full contact

Material: butyl-rubber

Body Protection

Impervious clothing, Flame retardant antistatic protective clothing. The type of protective equipment must be selected according to the concentration and amount of the dangerous substance at the specific workplace.

Respiratory protection

Where risk assessment shows air-purifying respirators are appropriate use a full-face respirator with multi-purpose combination (US) or type AXBEK (EN 14387) respirator cartridges as a backup to engineering controls. If the respirator is the sole means of protection, use a full-face supplied air respirator. Use respirators and components tested and approved under appropriate government standards such as NIOSH (US) or CEN (EU).

Control of environmental exposure

Prevent further leakage or spillage if safe to do so. Do not let product enter drains.

Use PPE that should be prepared during operation

• SECTION 9 Physical and Chemical Properties It mainly describes the appearance and physical and chemical properties of chemicals.

- Appearance and properties, PH, boiling point, melting point, relative density (water = 1), relative vapor density (air = 1), saturated vapor pressure, heat of combustion, critical temperature, critical pressure, octanol / water partition coefficient,
- Flash point, ignition temperature, explosion limit, solubility,
- Main uses and other special physical and chemical properties.
- Understand the physicochemical hazard characteristics of materials and control them when handling.

SECTION 9: Physical and chemical properties

9.1 Information on basic physical and chemical properties

Appearance Form: liquid, clear Colour: colourless Odour No data available Odour Threshold No data available

Melting point/freezing point

Melting point/range: -94 °C

Initial boiling point and

pН

d)

g)

Low flash

point,

flammabl

e liquid

Easy to

form an

explosiv

e mixture

after

leakage

56 °C at 1,013 hPa

No data available

boiling range Flash point

Evaporation rate

-17.0 °C - closed cup

No data available

Flammability (solid, gas) No data available i)

j) Upper/lower flammability or explosive limits

Upper explosion limit: 13 %(V) Lower explosion limit: 2 %(V)

k) Vapour pressure

533.3 hPa at 39.5 °C 245.3 hPa at 20.0 °C

I) Vapour density

No data available

Relative density

0.791 g/mL at 25 °C

Water solubility n)

completely miscible

log Pow: -0.24

Partition coefficient: noctanol/water





• SECTION 10 Stability and Reactivity Mainly describes the stability and reactivity of the chemical.

- Stability (whether it is decomposed into other harmful components when it is heated)
- Incompatible materials (reacting with other incompatible materials)
- Conditions to avoid (such as direct sunlight leading to high temperatures, etc.)
- Polymerization hazard,
- Decomposition products.



SECTION 10: Stability and reactivity

10.1 Reactivity

No data available

10.2 Chemical stability

Stable under recommended storage conditions.

10.3 Possibility of hazardous reactions

No data available

10.4 Conditions to avoid

Heat, flames and sparks

10.5 Incompatible materials

Bases, Oxidizing agents, Reducing agents, Acetone reacts violently with phosphorous oxychloride.

10.6 Hazardous decomposition products

Hazardous decomposition products formed under fire conditions. - Carbon oxides

Other decomposition products - No data available

In the event of fire: see section 5

Prevent all types of ignition sources during storage and operation

Incompatibility: Do not coexist with the above substances during storage, and avoid direct contact with them during operation.

| | 易燃物品 | 碱性及其他腐蚀物品 | 生有灰、漂白粉 其他 (无水肼、水合肼、氮水不得与氧化剂配存) 5材类、食用油脂 ^{5)、4)} | 23 24 25 26 | | 1 | × | Δ | Δ | Δ | Δ | × | X | | | | Δ | Δ | | | | | | × | Δ | | $\overline{}$ | 25 26 |
|------|-------------|---|---|----------------------------|-------------|---|---|---|-------------|--------|--------|-------------|---|-------------|----|------------|---------|----|----|----|----|---|---|---|----|----|---------------|----------|
| 相名務 | 366 | 酸性腐蚀品 | 溴 过氧化物 硝酸、发烟硝酸、硫酸、发烟碳酸、氯磷酸 其他酸性腐蚀物品 生石灰、漂白粉 | 19 20 21 22 23 | Δ | _ | × | \times \triangle \times \triangle | × Δ × | × | 1) | × | × | Δ | Δ | X Z X X | X A | Δ | Δ | _ | Δ | Δ | - | | 22 | 23 | | |
| 的科类和 | SESSIONAL | (H发孔剂不可与酸性 氰化物 其他毒害品 | 腐蚀物品及有毒或易燃酯类危险货物配存) | 16 17 18 | | Δ | Δ | Δ | × | Δ | Δ | × | | × | | × |).) | | 16 | 17 | 18 | | | | | | | |
| 危险品 | 過水燃烧 | 一級 物品(不得与含水液体 | *货物在同一库内配存》 | 13 14 15 | Δ. | × | × | A × | Δ | Δ | ΔΧ | × Δ × | Δ | Δ Δ × | _ | × / | 14 | 15 | ı | | | | | | | | | |
| 化学 | 19000253 | 不燃 | 不得与油脂类在同一库内配存) | 10 11 12 | Δ | × | × | × | × | Δ | Δ | × | Δ | 10 × | 11 | 2 | | | | | | | | | | | | |
| | 压缩气体和液化气 | 剧毒(液氯与液氯不 | 能在一库内配存) | 8 | Δ. | × | X | × | × | × | 350 | 8 | 9 | | | | | | | | | | | | | | | |
| | 氧 化 剂 | 有机氧化剂 亚硝酸盐、亚氯酸盐 其他无机氧化物 ²⁾ | 、次亚氯酸盐2) | 5 6 | Δ | Δ | Δ | × A | 5 × | 6 × | 5 | | | | | | | | | | | | | | | | | |
| | 爆 炸 品 | 点火器材 起爆器材 炸药和爆炸性物品(其他爆炸品 | 不同品名的不得在同一库内储存) | 1 2 3 4 | 1 × △ | _ | | 4 | | | | | | | | | | | | | | | | | | | | |
| 化 | 学危险品的 |)种类和名称 | | 配存順号 | | | | | | | | | | | | | | | | | | | | | | | | |

• SECTION 11 Toxicological Information Provides toxicological information on chemicals.

- Acute toxicity of different contact methods (LD50)
- Irritant, allergenic, subacute and chronic toxicity, mutagenicity, teratogenicity, carcinogenicity, etc.





SECTION 11: Toxicological information

11.1 Information on toxicological effects

Acute toxicity

LD50 Oral - Rat - 5,800 mg/kg

Remarks: Behavioral:Altered sleep time (including change in righting reflex). Behavioral:Tremor. Behavioral:Headache. Ingestion may cause gastrointestinal irritation, nausea, vomiting and diarrhoea.

LC50 Inhalation - Rat - 8 h - 50,100 mg/m3

Remarks: Drowsiness Dizziness Unconsciousness

LD50 Dermal - Guinea pig - 7,426 mg/kg

Skin corrosion/irritation

Skin - Rabbit

Result: Mild skin irritation - 24 h

Serious eye damage/eye irritation

Eyes - Rabbit

Result: Eye irritation - 24 h

Respiratory or skin sensitisation

- Guinea pig

Result: Does not cause skin sensitisation.

Germ cell mutagenicity

No data available

Carcinogenicity

This product is or contains a component that is not classifiable as to its carcinogenicity based on its IARC, ACGIH, NTP, or EPA classification.

IARC: No component of this product present at levels greater than or equal to 0.1% is identified as

probable, possible or confirmed human carcinogen by IARC.

Reproductive toxicity

No data available

Specific target organ toxicity - single exposure

May cause drowsiness or dizziness.

Specific target organ toxicity - repeated exposure

No data available

• SECTION 12 Ecological Information It mainly states the environmental and ecological effects, behaviors and outcomes of chemicals.

- Biological effects (eg LD50),
- Biodegradability,
- Bioaccumulation,
- Environmental migration and other harmful environmental impacts.
- Understand the seriousness of the short-term and long-term damage to the surrounding environment (water, atmosphere, soil)

SECTION 12: Ecological information

12.1 Toxicity

Toxicity to fish LC50 - Oncorhynchus mykiss (rainbow trout) - 5,540 mg/l - 96 h

Toxicity to daphnia and

LC50 - Daphnia magna (Water flea) - 8,800 mg/l - 48 h

other aquatic invertebrates

Toxicity to algae Remarks: No data available

12.2 Persistence and degradability

Biodegradability Result: 91 % - Readily biodegradable.

(OECD Test Guideline 301B)

12.3 Bioaccumulative potential

Does not bioaccumulate.

12.4 Mobility in soil

No data available

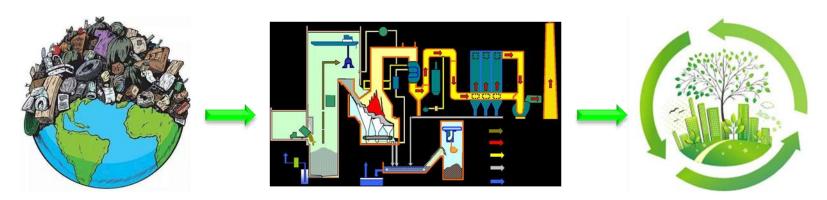
12.5 Results of PBT and vPvB assessment

This substance/mixture contains no components considered to be either persistent, bioaccumulative and toxic (PBT), or very persistent and very bioaccumulative (vPvB) at levels of 0.1% or higher.

12.6 Other adverse effects

No data available

- SECTION 13 Disposal Considerations Refers to the safe handling of packaging contaminated with chemicals and non-useable chemicals.
- include: Disposal methods and precautions.



SECTION 13: Disposal considerations

13.1 Waste treatment methods

Product

Offer surplus and non-recyclable solutions to a licensed disposal company. Waste material must be disposed of in accordance with the Directive on waste 2008/98/EC as well as other national and local regulations. Leave chemicals in original containers. No mixing with other waste. Handle uncleaned containers like the product itself.

Contaminated packaging

Dispose of as unused product.



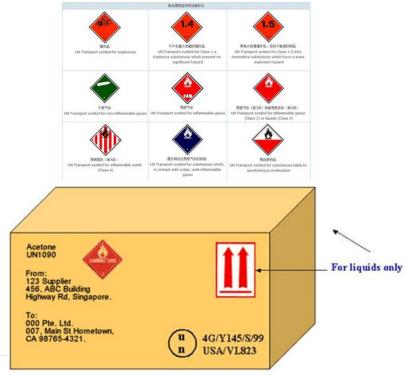
Tray or pallet for drums, label missing



Inner cap missing

• SECTION 14 Transport Information Mainly refers to the classification and numbering of domestic and international chemical packaging and transportation requirements and transportation regulations.

- Dangerous goods number,
- Packing category,
- Packaging mark,
- Method of packing,
- UN number and transportation precautions.



SECTION 14: Transport information 14.1 UN number ADR/RID: 1090 IMDG: 1090 IATA: 1090 14.2 UN proper shipping name ADR/RID: ACETONE IMDG: ACETONE IATA: Acetone 14.3 Transport hazard class(es) ADR/RID: 3 IMDG: 3 IATA: 3 14.4 Packaging group ADR/RID: II IMDG: II IATA: II 14.5 Environmental hazards ADR/RID: no IATA: no IMDG Marine pollutant: no 14.6 Special precautions for user No data available

- SECTION 15 Regulatory Information Mainly refers to the legal provisions and standards for chemicals management.
- Such as :
 - Globally Harmonized System of Classification and Labeling of Chemicals, Abbreviation GHS
 - REACH Regulation (1907/2006)
 - CLP Regulation (EC NO. 1272/2008)
 - 67/548/EEC appendix III
 - Instruction 2001/58/EC
 - Material Classification and Label Management Directive 67/548/EEC (ie DSD Directive)
 - Classification labeling instructions for mixtures 1999/45/EEC (ie DPD directives)
 - General Rules for the Classification and Hazard of Chemicals (GB-13690-2009)
 - Preparation of Chemical Safety Data Sheet (GB 16483-2008)
 - Chemical Safety Labeling Regulations (GB 15258-2009)



15. REGULATORY INFORMATION

SARA 302 Components

No chemicals in this material are subject to the reporting requirements of SARA Title III, Section 302.

SARA 313 Components

This material does not contain any chemical components with known CAS numbers that exceed the threshold (De Minimis) reporting levels established by SARA Title III, Section 313.

SARA 311/312 Hazards

Fire Hazard, Acute Health Hazard, Chronic Health Hazard

Massachusetts Right To Know Components

| | CAS-No. | Revision Date |
|--|---------|---------------|
| Acetone | 67-64-1 | |
| Danuardiania Dielat Ta Kuassi Cammananta | | |

Pennsylvania Right To Know Components

| Acetone | 67-64-1 | r to note in Date |
|-------------------------------------|---------|-------------------|
| New Jersey Right To Know Components | CAS-No | Dovision Data |
| Acetone | 67-64-1 | Revision Date |

California Prop. 65 Components

This product does not contain any chemicals known to State of California to cause cancer, birth defects, or any other reproductive harm.

CAS-No.

Revision Date

• SECTION 16 Other Information Mainly provides other information that is important to safety.

- References,
- Time of the form (SDS),
- Department filling in the form,
- Data review unit, etc.

16. OTHER INFORMATION

Full text of H-Statements referred to under sections 2 and 3.

H225 Highly flammable liquid and vapour. H319 Causes serious eye irritation.

H336 May cause drowsiness or dizziness.

HMIS Rating

Health hazard: 2
Chronic Health Hazard: *
Flammability: 3
Physical Hazard 0

NFPA Rating

Health hazard: 2 Fire Hazard: 3 Reactivity Hazard: 0

Further information

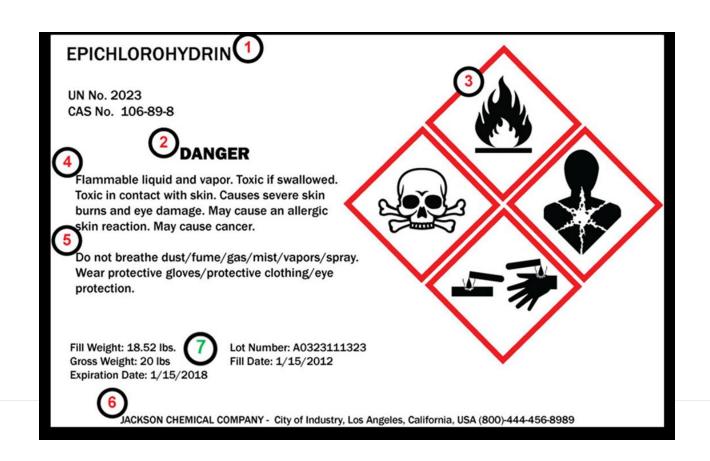
Copyright 2016 Sigma-Aldrich Co. LLC. License granted to make unlimited paper copies for internal use only. The above information is believed to be correct but does not purport to be all inclusive and shall be used only as a guide. The information in this document is based on the present state of our knowledge and is applicable to the product with regard to appropriate safety precautions. It does not represent any guarantee of the properties of the product. Sigma-Aldrich Corporation and its Affiliates shall not be held liable for any damage resulting from handling or from contact with the above product. See www.sigma-aldrich.com and/or the reverse side of invoice or packing slip for additional terms and conditions of sale.

Preparation Information

Sigma-Aldrich Corporation Product Safety – Americas Region 1-800-521-8956

Version: 6.1 Revision Date: 05/28/2017 Print Date: 10/14/2019

- safety labels
- Safety labels use a combination of words, graphic symbols and codes to indicate the hazards and safety precautions of the chemical.



SDS and

1 化学品名称 A组分: 40%; B组分: 60%

危险







极易燃液体和蒸气,食入致死,对水生生物毒性非常大

【预防措施】

- 远离热源、火花、明火、热表面。使用不产生火花的工具作业。
- 保持容器密闭。
- 采取防止静电措施,容器和接收设备接地、连接。
- 使用防爆电器、通风、照明及其他设备。
- 或防护手套、防护眼镜、防护面罩。
- 操作后彻底清洗身体接触部位。
- 作业场所不得进食、饮水或吸烟。
- 禁止排入环境。

【事故响应】

- 如皮肤(或头发)接触;立即脱掉所有被污染的衣服。用水冲洗皮肤、淋浴。
- 食入: 催吐, 立即就医。
- 收集泄漏物。
- 火灾时,使用干粉、泡沫、二氧化碳灭火。

【安全储存】

- 在阴凉、通风良好处储存。
- 上锁保管。

【废弃处置】

本品或其容器采用焚烧法处置。

请参阅化学品安全技术说明书

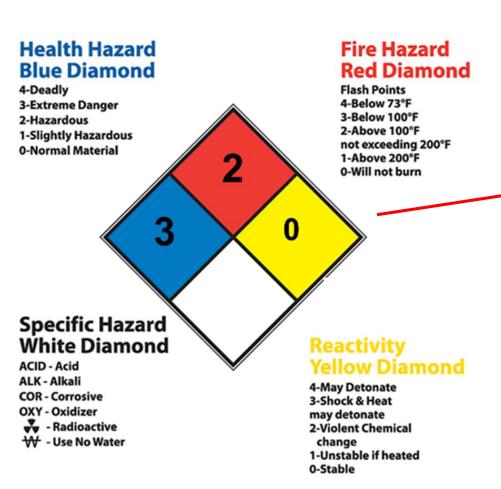
邮编: XXXXXX

巾话: XXXXXX

化学事故应急咨询电话: ××××××

SDS and safety labels - GHS pictograms

| Flame | Flame over circle | Exploding bomb |
|------------------|-------------------|----------------------|
| | | 3 |
| Corrosion | Gas cylinder | Skull and crossbones |
| | | |
| Exclamation mark | Environment | Health Hazard |
| | *** | |





• Problem: Chemical reagents' labeling is unclear or missing.

- The role of SDS and safety labels
- According to the content of SDS or safety label, the requirements for storage and use of chemicals can be clarified;
- It can be used as the basis for matching PPE;
- Section 5, 6, and 8 of the SDS can be used as emergency measures.

| 5. Fire Fighting Measures | Take the school chemical warehouse storage as an example: Storage needs to be based on the environmental sensitivity of |
|-------------------------------------|---|
| ID ALL IIIEIIIAI KEIEASE WIEASIIIES | each substance (sunlight, ambient temperature), compatibility with other chemicals, fire-fighting characteristics; |
| | How to control the amount of leakage regarding varied packaging; |
| p 1 = 11 / 510 001 00110 | Extinguishing media when the material is on fire; |
| 14 0 0 1 111 1 1 1 1 1 1 | Design the maximum storage and storage location for storage based on the nature of existing chemicals. |



For on-site operation:

• SDS information facilitates on-site development of "Occupational Hazard Notification Card"

The information includes:

- GHS pictogram;
- Health hazard;
- Physical and chemical properties;
- Emergency treatment measures;
- PPE.





- How to use SDS to assist in emergency response under abnormal conditions?
- How to ensure the safety of evacuated personnel;
- How to ensure the safety of rescuers, prevent injuries during rescuing action and how to clean up chemicals on rescuers after completing the rescue;
- How to ensure chemical leaks, fires, occupational injuries are effectively handled;
- How to ensure the environment (not flowing into the rainwater system, not entering the soil environment, not dissipate into the atmosphere)

Chemicals management system

---Informatization, scientific management





"Nothing we do is worth getting hurt for !" 沒什么是值得以牺牲安全作为代价!